

Name _____ Date _____ Class _____

12-4 Review and Reinforcement

Calorimetry

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

water	heat capacity	specific heat
heat	calorimetry	temperature
oxygen	calorimeter	carbohydrates

- The study of heat flow and heat measurement is called _____.
- The words calorimeter, calorimetry, and calorie are all derived from the Latin word *calor*, which means _____.
- The amount of heat needed to raise an object's temperature depends on its _____.
- Every substance has a _____, which tells you how much heat is necessary to raise the temperature of 1 gram of the substance by 1 Celsius degree.
- _____ has one of the highest specific heats of any common substance.
- A transfer of heat is detected by measuring a _____ change.
- A _____ is a well-insulated container used to measure temperature changes.
- On the average, _____ supply 17 kJ/g (4 Cal/g) of energy.
- Foods are reacted with _____ in a laboratory calorimeter to determine their energy values.

Use your knowledge from Section 12-4 to write the meaning of each of the following symbols.

- T_i _____
- T_f _____
- q_{rxn} _____
- q_{sur} _____
- m _____
- C _____

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4 Review And Reinforcement Calorimetry Answers

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3-4 Review and Reinforcement (continued) Complete each of the following nuclear reactions by filling in the blank with the correct particle. 11. $^{226}\text{Ra} \rightarrow ^{222}\text{Rn} + 88\text{He} + 2\text{e}^-$ 12. 13. Write the nuclear equations for the following reactions. 14. alpha decay of polonium-210. $^{210}\text{Po} \rightarrow ^{206}\text{Pb} + ^4\text{He}$ 15. beta decay of tritium (^3H) 16. alpha decay of thorium-230 (^{230}Th) On the line at the left, write the letter of the ...

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